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SYNTHESIS AND STRUCTURE OF AROMATIC COMPOUNDS CARRYING TWO 1-ADAMANTYLS ON ADJACENT POSITIONS: 3,4-DI-1-ADAMANTYLTHIOPHENE, o-DI-1-ADAMANTYLBENZENE, AND 4,5-DI-1-ADAMANTYLPYRIDAZINE

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<u>Abstract</u> 3,4-Di-1-adamantylthiophene, *o*-di-1-adamantylbenzene, and 4,5-di-1-adamantylpyridazine, which carry two bulky 1-adamantyl groups on adjacent positions, were satisfactorily synthesized starting from 1,5-di-1-adamantyl-3-thiapentane-1,5-dione. Synthesis, properties, and single crystal structure analysis of these compounds are described.

1-Adamantyl is a very bulky substituent similar to *tert*-butyl. It can be considered a kind of "tied-back" *tert*-butyl group but is far less flexible and thus might behave as a bulkier substituent than *tert*-butyl. To our knowledge, no report has appeared on the successful synthesis of five- or six-membered aromatic rings carrying two 1-adamantyl groups on adjacent positions. These would be sterically more strained than the corresponding di-*tert*-butyl-substituted compounds. We report here the synthesis and structure of such molecules, 3,4-di-1-adamantylthiophene (3), *o*-di-1-adamantyl benzene (5), and 4,5-di-1-adamantyl pyridazine (8).

Diketo sulfide 1 is easily obtainable² by reaction of sodium sulfide with

commercially available 1-adamantyl bromomethyl ketone. Intramolecular pinacol reduction of 1 by a low valent titanium reagent, prepared from titanium(IV) chloride and zinc powder, at -18 °C for 9 h in THF afforded the diol 2 in 34% yield. 3 p-Toluenesulfonic acid catalyzed dehydration of 2 in refluxing benzene for 0.5 h produced the expected thiophene 3 in 60% yield. 3a,c Oxidation of 3 with m-chloroperbenzoic acid (MCPBA) in dichloromethane at room temperature gave the corresponding 1,1-dioxide 4 in 75% yield. 3c

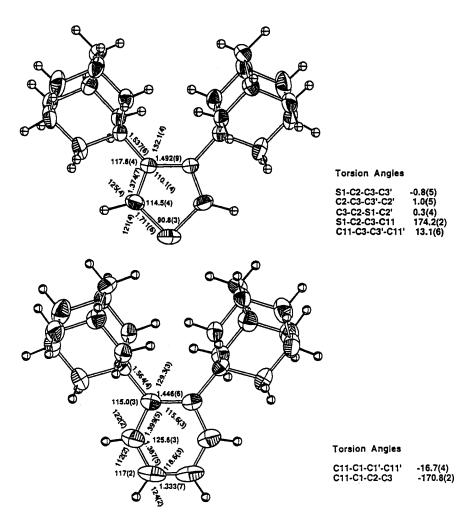
The Diels-Alder reaction of dioixde **4** with phenyl vinyl sulfone (3 equiv) in refluxing *o*-dichlorobenzene for 24 h afforded *o*-di-1-adamantylbenzene (5) in 92% yield with loss of sulfur dioxide and benzenesulfinic acid from the initial adduct.⁴ Similarly, the reaction of **4** with dimethyl acetylenedicarboxylate (DMAD) in refluxing *o*-dichlorobenzene for 7 h furnished dimethyl 4,5-di-1-adamantylphthalate (6) in 90% yield with extrusion of sulfur dioxide from the initial adduct.⁴

The reaction of 4 with excess 4-phenyl-1,2,4-triazoline-3,5-dione (7.5 equiv) in refluxing toluene afforded the bis-adduct 7 in 72% yield. Treatment of 7 with potassium hydroxide in methanol at room temperature followed by air oxidation and nitrogen extrusion led directly to 4,5-di-1-adamantylpyridazine (8) in 82% yield (Scheme 1).⁵

The two 1-adamantyl substituents of **3**, **5**, and **8** are equivalent in the ¹H NMR spectra, which confirms that rotation about the bond from the aromatic ring to adamantyl is fast on the ¹H NMR time scale at room temperature. A similar conclusion is reached from ¹³C NMR analysis.

Treatment of **3** with aluminum chloride in carbon disulfide at room temperature for 6 days brought about quantitative isomerization to 2,4-di-1-adamantylthiophene. Treatment of **5** under similar conditions resulted in exhaustive deadamantylation, although the use of benzene as the solvent gave 1-adamantylbenzene in 41% yield.

ORTEP drawings of compounds **3** and **5** are given below with relevant bond lengths and angles. The C(3)-C(4) bond of **3** is longer than that of the parent thiophene by *ca.* 0.07 Å and the adamantyl-C(3)-C(4) bond angle is larger than the H-C(3)-C(4) bond angle of the parent thiophene by *ca.* 9°. Two adamantyls are twisted with a torsion angle of 13.1°. In compound **5** the adamantyl-C(1)-C(2) bond angle is 129.3° and *two adamantyls are twisted with a torsion angle of 16.6°, thus making the benzene ring nonplanar!* The C(1)-C(2) bond of **5** is as long as 1.446 Å and the C(4)-C(5) bond as short as 1.333 Å.



ORTEP Drawings of 3,4-Di-1-adamantylthiophene (top) and o-Di-1-adamantylbenzene (bottom)

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